

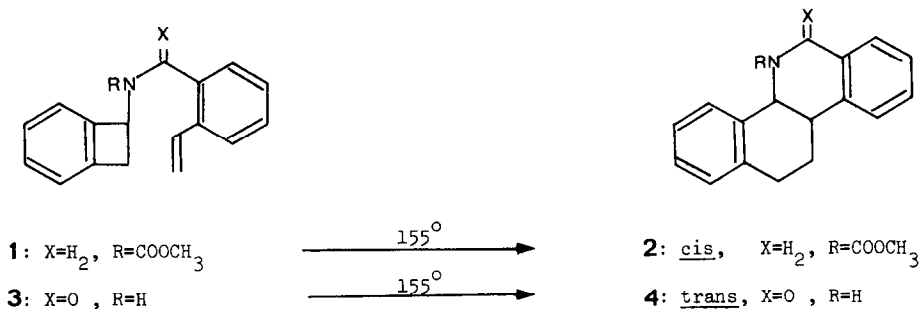
STERIC CONTROL OF INTRAMOLECULAR ORTHO-QUINODIMETHANE-CYCLOADDITIONS

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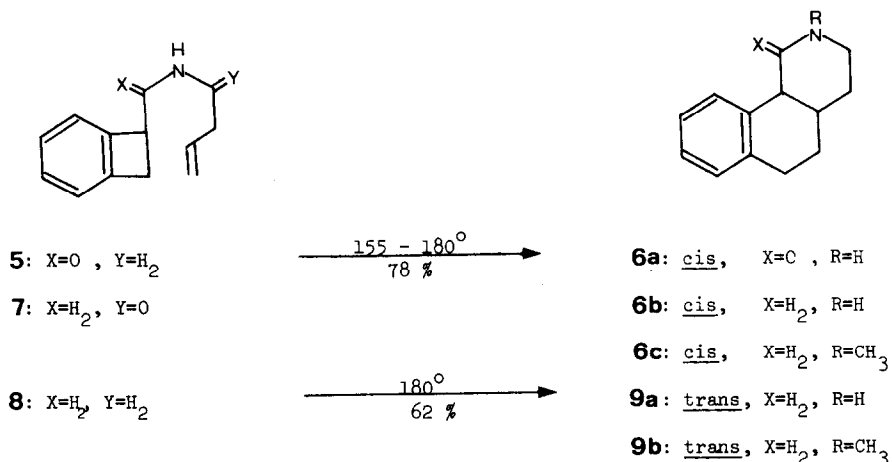
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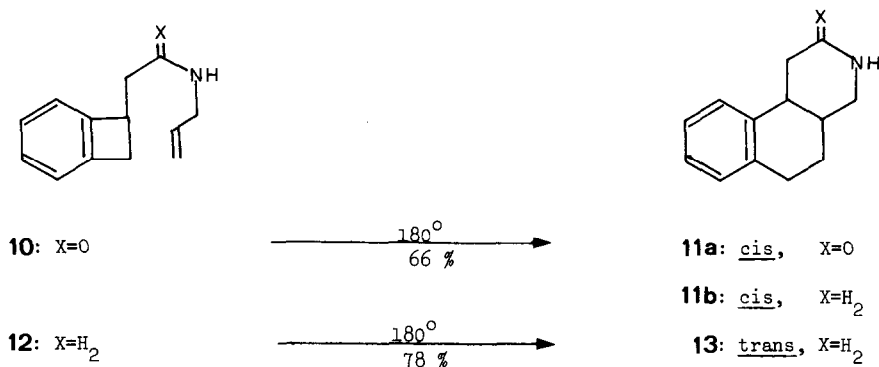
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Endo-adducts normally predominate in the cycloaddition of non-conjugated dienophiles (1) although nonbonding interactions of substituents may force the reaction towards exo-adducts (2). A simple and effective control of the endo-exo-ratio has been exemplified recently by the intramolecular cycloadditions **1** \longrightarrow **2** and **3** \longrightarrow **4** (3).



Similar syntheses of either cis- or trans-fused ring systems from non-conjugated dienophiles were examined as follows.





Thermolysis of the butenylamide **5** in refluxing dichlorobenzene gave in 78 % yield the cis-fused lactam **6a** (4), which was reduced with aluminium hydride to the amine **6b** (5) (hydrogen maleinate, m.p. 182 - 183° C). However, after the butenylamine **8** (5) was refluxed in dichlorobenzene for 16 h, the trans-fused benz(h)isoquinoline **9a** (5) (hydrogen maleinate, m.p. 183 - 184° C) was isolated in 62 % yield. Under identical conditions the allylamide **10** (5, 6) afforded the cis-fused benz(f)isoquinoline **11a** (5, 8, 9), (m.p. 189 - 191° C, 66 % yield), whereas the amine **12** (5) gave the trans-product **13** (5) (m.p. 79 - 80° C, 78 % yield). A g.l.c. analysis of the crude pyrolysates (10) is outlined in Table 1.

Table 1. G.l.c. analysis of stereoisomeric octahydrobenz-isoquinolines from 1-substituted benzocyclobutanes

starting material	reaction conditions	products	
		cis	trans
5	a,b	83 % 6b	16 % 9a
7	a,b	64 % 6b	31 % 9a
8	a	27 % 6b	72 % 9a
10	a,b	79 % 11b	20 % 13
12	a	12 % 11b	87 % 13

a) 16 h/180° C; b) AlH₃/THF

According to g.l.c. analysis (10), the isolated products **6b**, **9a**, **11a**, **11b** and **13** remained unchanged when refluxed in dichlorobenzene for 16 h, whereas equilibration of the isomeric methylamines **6c** (5) and **9b** (5) with 5 equiv. of potassium tert-butyrate in dimethylsulfoxide at 60° C for 16 h led to a 2:3-mixture of cis-/trans-isomers. Hence it appears, that the stereochemistry of the reactions of the benzocyclobutenes **8**, **10** and **12** is controlled kinetically. Also the thermolysis of the amide **7** most probably is subject to kinetic control (11). Taking into account the intermediacy of trans-quinodimethanes (12) it follows that the

thermal reorganization of the amides **7** and **10** passes predominantly through the endo-transition state I, whereas the exo orientation II is favoured on cyclization of the amines **8** and **12**.



These results therefore illustrate the general possibility to direct the steric course of intramolecular cycloadditions by conformational variations of the bridge between the dipole and the dipolarophile.

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- [5] Elemental analytical data, ir- and pmr-spectra in agreement with the assigned structures were obtained for all new compounds.
- [6] The amide **10**, m.p. 44 - 46° C, was prepared in 81 % yield by reaction of benzocyclobutene-1-acetyl chloride, c.f. Ref. [7], with an excess of allylamine in dichloromethane.
- [7] C.D. Gutsche, G.L. Bachmann and R.S. Coffey, *Tetrahedron* **18**, 617 (1962)
- [8] The structure of **11a** was deduced from pmr-decoupling experiments (CDCl₃ + 0.5 equiv. Eu(DPM)₃) kindly carried out by H.R. Loosli, Sandoz Ltd.
- [9] For a previous approach to 1,2,3,4,4a,5,6,10b-octahydrobenz(f)isoquinolines see: R.J.P. Barendse, W.N. Speckamp and H.O. Huisman, *Tetrahedron Letters* **1970**, 5301.

- [10] For g.l.c.-analysis (capillary column OV 225/185°) the amines were acylated with trifluoroacetic acid anhydride/pyridine in dichloromethane.
- [11] Heating the amide **7** in boiling *o*-dichlorobenzene for 0.5 h, followed by reduction with AlH₃, gave the stereo-isomers **6b** and **9a** in yields of 0.84 % and 0.53 %, respectively (according to g.l.c. analysis). However, thermolysis of the amide **5** in refluxing bromobenzene for 5 h and subsequent reduction led to the epimers **6b** and **9a** in yields of 9 % and 11 % respectively, which indicates that the stereo-selective formation of the lactam **6a** in refluxing *o*-dichlorobenzene is at least partly the result of thermodynamic control.
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